

Available online at www.sciencedirect.com



Tetrahedron Letters

Tetrahedron Letters 48 (2007) 7173-7176

## Stereoselective synthesis of the phytotoxic nonenolide herbarumin-I from L-ascorbic acid

K. Nagaiah,\* D. Sreenu, R. Srinivasa Rao and J. S. Yadav

Division of Organic Chemistry, Indian Institute of Chemical Technology, Hyderabad 500 007, India

Received 31 May 2007; revised 23 July 2007; accepted 30 July 2007 Available online 2 August 2007

Abstract—A stereoselective synthesis of herbarumin-I in 22% overall yield, starting from L-ascorbic acid derived (S)-2,3-O-isopropylidine glyceraldehyde as a chiral template is reported. Stereoselective allylation and vinylation to control the required stereogenic centres and macrolactonisation followed by a ring-closing metathesis (RCM) are the key steps. © 2007 Published by Elsevier Ltd.

Bioassay guided fractionation of a culture broth and mycelium of the fungus Phoma herbarum Westend (Sphaeropsidaceae) led to the discovery of three novel nonenolides named herbarumin-I (1), II (2) and III (3) (Fig. 1).<sup>1</sup> These lactones exhibited significant phytotoxic effects when tested against the seedlings of Amaranthus hypochondriacus at very low concentrations using a Petri dish bioassay.<sup>2</sup> Amongst these lactones herbarumin-I (1) shows promising phytotoxic effects with IC<sub>50</sub> values as low as  $5.43 \times 10^{-5}$ . Enzyme-inhibition studies of compounds 1-3 also suggested an interesting behaviour superior to chlorpromazine, as calmodulin-dependent enzyme cyclic nucleotide (cAMP) phosphodiesterase calmodulin inhibitors without interfering with the basal activity or the independent form of the enzyme.<sup>1a,3</sup> As a result of these activities together with the fact that

closely related compounds such as pinolidoxin  $(4)^4$  and lethaloxin  $(5)^5$  (Fig. 1) also exert significant phytotoxicity, makes this class of compounds promising new lead structures in the search for novel herbicidal agents.

Several approaches towards the asymmetric synthesis of these ten-membered lactones 1–3 have appeared in the literature.<sup>6</sup> The asymmetric synthesis of these lactones involves ring-closing metathesis (RCM) or Nozaki– Hiyama–Kishi reactions as the key steps for the formation of the ten-membered ring starting from various sugars, such as D-ribose, L-arabinose and D-glucose as chiral-pool templates. Despite the superior biological activity of herbarumin-I (1) compared to herbarumin-II (2) and herbarumin-III (3), only two asymmetric syntheses of 1 have been reported in the literature.<sup>6a,b,d</sup>



Figure 1.

Keywords: L-Ascorbic acid; Allylation; Vinylation; Macrolactonisation; Ring-closing metathesis.

<sup>\*</sup> Corresponding author. Tel.: +91 40 27193154; fax: +91 40 27160512; e-mail: nagaiah@iict.res.in

<sup>0040-4039/\$ -</sup> see front matter @ 2007 Published by Elsevier Ltd. doi:10.1016/j.tetlet.2007.07.191



Scheme 1. Retrosynthesis of herbarumin-I (1).

Herein, we describe the stereoselective synthesis of herbarumin-I (1) starting from L-ascorbic acid derived (S)-2,3-O-isopropylidine glyceraldehyde as a chiral template. Enantioselective allylation and vinylation reactions to control the required stereogenic centres and macrolactonisation followed by a ring-closing metathesis (RCM) are key-steps. A retrosynthetic strategy for 1 is outlined in Scheme 1.

Initially we investigated the Zn-mediated<sup>7</sup> stereoselective allylation of (S)-2,3-O-isopropylidine glyceraldehyde  $2^8$ 

to furnish the corresponding homoallylic alcohols 3a and 3b as a mixture of diastereomers (*anti:syn* = 95:5) in 92% overall yield which was separated by silica gel chromatography. The major diastereomer 3a had the required stereochemistry at C-3. Homoallylic alcohol 3a on hydrogenation using 10% Pd/C and subsequent PMB protection of the hydroxy group with PMBBr and NaH at 0 °C afforded **5**. Hydrolysis of the 2,3-*O*-isopropylidene moiety of **5** with 1 N HCl in THF at ambient temperature afforded diol **6**, which on TES-protection with 2.2 equiv of TESCl and imidazole in DCM



Scheme 2. Reagents and conditions: (a) Zn, allyl bromide, THF, satd NH<sub>4</sub>Cl, 0 °C to rt, 4 h, 92%; (b) Pd/C, H<sub>2</sub>, EtOH, rt, 4 h, 100%; (c) PMBBr, NaH, THF, 0 °C to rt, 4 h, 85%; (d) 1 N HCl, THF, rt, 6 h, 95%; (e) TESCl, imidazole, DCM, 0 °C to rt, 3 h, 85%; (f) (COCl)<sub>2</sub>, dry DMSO, Et<sub>3</sub>N, DCM, -78 °C, 3 h, 80%; (g) vinylmagnesium bromide, dry THF, 0 °C to rt, 3 h, 90%; (h) 1 N HCl, THF, rt, 3 h, 92%; (i) 2,2-DMP, cat. PPTS, rt, 6 h, 95%; (j) DDQ, DCM-H<sub>2</sub>O (19:1), rt, 1 h, 92%; (k) 5-hexenoic acid, 2,4,6-trichlorobenzoyl chloride, DMAP, THF, 12 in THF, 0 °C to rt, 4 h, 85%.



Scheme 3. Reagents and conditions: (a)  $30 \mod \%$  Ru-II, CH<sub>2</sub>Cl<sub>2</sub>, reflux, 12 h, (85% Z-isomer); (b)  $30 \mod \%$  Ru-I, CH<sub>2</sub>Cl<sub>2</sub>, reflux, 8 h, (*E*:*Z* = 80:20 = 82%); (c) 1 N HCl, THF, 50 °C, 4 h, 87%.

provided di-TES compound 7. Compound 7 on Swern oxidation<sup>9</sup> furnished aldehyde 8 via domino deprotection of the primary O-TES group and subsequent oxidation of the primary alcohol in 80% yield. Stereoselective vinylation of aldehyde 8 with in situ generated vinylmagnesium bromide afforded the corresponding allylic alcohols as a mixture of diastereomers 9a and 9b (anti: syn = 90:10) in 90% overall yield and these diastereomers were separated by silica gel chromatography. The major diastereomer 9a had the required stereochemistry at C-3. Compound 9a on treatment with 1 N HCl gave diol 10, which on reaction with 2,2-DMP in the presence of PPTS furnished acetonide 11. PMB deprotection of 11 with DDO yielded intermediate 12, whose spectral and optical rotation values were in good agreement with the literature.<sup>6a,b</sup> Esterification of the hydroxyl group of 12 with 5-hexenoic acid<sup>10</sup> under Yamaguchi conditions<sup>11</sup> at room temperature afforded 13 in 85% yield, which has all the structural requirements as well as sense of chirality for the ring-closing metathesis (Scheme 2).

The RCM of 13 in the presence of Grubbs' secondgeneration catalyst led to the exclusive formation of the undesired (Z)-isomer 14b in 85% yield, whose spectral data compared well with that reported in the literature.<sup>6b</sup> However, compound 13, in the presence of Grubbs' first generation catalyst yielded an E/Z mixture of cyclic olefins (14a:14b = E:Z = 80:20) in 82% overall yield.<sup>12</sup> The diastereomers were separated by chromatography using an AgNO<sub>3</sub>-silica gel column. Cleavage of the acetal group in diastereomer 14a under acidic conditions afforded herbarumin-I (1),<sup>13</sup> whose spectral and analytical data were consistent with those reported in the literature (Scheme 3).<sup>1,6b</sup>

In summary, we have developed a simple, convenient and efficient approach for the synthesis of herbarumin-I involving a sequence of reactions starting from (S)-2,3-O-isopropylidene glyceraldehyde. This approach offers a high overall yield, useful stereoselectivity and readily available starting materials at low cost and involves simple experimental conditions, which makes it a useful and attractive process for the total synthesis of herbarumin-I.

## Acknowledgement

D.S. and R.S.R. thank CSIR, New Delhi, for the award of fellowships.

## **References and notes**

- (a) Rivero-Cruz, J. F.; Macias, M.; Cerda-Garcia-Rojas, C. M.; Mata, R. J. Nat. Prod. 2003, 66, 511–514; (b) Rivero-Cruz, J. F.; Garcia-Aguirre, G.; Gerda-Garcia-Rojas, C.; Mata, R. Tetrahedron 2000, 56, 5337–5344.
- Mata, R.; Macias, M.; Rojas, S.; Lotina-Hensen, B.; Toscano, R.; Anaya, A. *Phytochemistry* 1998, 49, 441–449.
- (a) Leung, P. C.; Taylor, W. A.; Wang, J. H.; Tripton, C. L. J. Biol. Chem. 1984, 259, 2742–2747; (b) Macias, M.; Ulloa, M.; Gamboa, A.; Mata, R. J. Nat. Prod. 2000, 63, 757–761.
- (a) Evidente, A.; Capasso, R.; Abouzeid, M. A.; Lanzetta, R.; Vurro, M.; Bottalico, A. *J. Nat. Prod.* **1993**, *56*, 1937– 1943; (b) Evidente, A.; Lanzetta, R.; Capasso, R.; Vurro, M.; Bottalico, A. *Phytochemistry* **1993**, *34*, 999–1003.
- (a) Arnone, A.; Nasini, G.; Merlini, L.; Ragg, E.; Assante, G. J. Chem. Soc., Perkin Trans. 1 1993, 145–151; (b) Drager, G.; Kirschning, A.; Thiericke, R.; Zerlin, M. Nat. Prod. Rep. 1996, 13, 365–375.
- (a) Furstner, A.; Radkowski, K. Chem. Commun. 2001, 671–672;
  (b) Furstner, A.; Radkowski, K.; Wirtz, C.; Goddard, R.; Lehmann, C. W.; Mynott, R. J. Am. Chem. Soc. 2002, 124, 7061–7069;
  (c) Diez, E.; Dixon, D. J.; Ley, S. V.; Polara, A.; Rodriguez, F. Helv. Chim. Acta 2003, 86, 3717–3729;
  (d) Sabino, A. A.; Pilli, R. A. Tetrahedron Lett. 2002, 43, 2819–2822;
  (e) Gurjar, M. K.; Karmakar, S.; Mohapatra, D. K. Tetrahedron Lett. 2004, 45, 4525– 4526;
  (f) Nanda, S. Tetrahedron Lett. 2005, 46, 3661–3663;
  (g) Gurjar, M. K.; Nagaprasad, R.; Ramana, C. V.; Mohapatra, D. K. Arkivoc 2005, 3, 237–257;
  (h) Salaskar, A.; Sharma, A.; Chattopadhyay, S. Tetrahedron: Asymmetry 2006, 17, 325–329.
- (a) Marton, D.; Stivanello, D.; Tagliavini, G. J. Org. Chem. 1996, 61, 2731–2737; (b) Christian, P.; Jean, L. L. J. Org. Chem. 1985, 50, 910–912; (c) Petrier, C.; Einhorn, J.; Luche, J. L. Tetrahedron Lett. 1985, 26, 1449–1452; (d) Cathy, E.; Jean, L. L. J. Organomet. Chem. 1987, 322, 177–183; (e) Yong, X.; Glenn, D. P. J. Org. Chem. 2002, 67, 7158–7161; (f) Jackson, D. Y. Synth. Commun. 1988,

18, 337–341; (g) Janusz, J.; Stanislaw, P.; Tomasz, B. *Tetrahedron* **1986**, *42*, 447–448.

- (a) Jung, M. E.; Shaw, T. J. J. Am. Chem. Soc. 1980, 102, 6304–6311; (b) Abushnab, E.; Venishetti, P.; Leiby, R. W.; Singh, H. K.; Mikkilineni, A. B.; Wu, D. C.-J.; Saibaba, R.; Panzica, P. J. Org. Chem. 1988, 53, 2598–2602.
- 9. Rodriguez, A.; Nomen, M.; Spur, B. W.; Godfroid, J. J. *Tetrahedron Lett.* **1999**, 40, 5161–5164.
- 10. Krapcho, A. P. Synthesis 1982, 805-822, and 893-914.
- Jnanaga, J.; Hirata, K.; Saeki, H.; Katsuki, T.; Yamaguchi, M. Bull. Chem. Soc. Jpn. 1979, 52, 1989–1993.
- Fortanet, J. G-.; Murga, J.; Falomir, E.; Carda, M.; Marco, J. A. . J. Org. Chem. 2005, 70, 9822–9827.
- 13. Spectral data for selected products: 1-(2,2-Dimethyl-[1,3]dioxolan-4-yl)-but-3-en-1-ol (**3a**): Pale yellow oil;  $[\alpha]_D^{25} -5.40$  (c 0.5, MeOH); IR (neat):  $v_{max}$ : 3454 (br, OH), 2986, 1375, 1214, 1064 cm<sup>-1</sup>; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>);  $\delta$  5.90–5.70 (m, 1H), 5.20–5.00 (m, 2H), 4.00– 3.90 (m, 3H), 3.75–3.65 (m, 1H), 2.40–2.20 (m, 2H), 1.40–1.30 (d, 6H); <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>): 134.0, 118.0, 109.0, 88.0, 71.0, 65.2, 37.5, 27.8, 25.2; MS-EIMS: m/z (%): 172 (M<sup>+</sup>, 5), 157 (10), 141 (13), 101 (70), 59 (35), 43 (100); HRMS calcd for C<sub>9</sub>H<sub>16</sub>O<sub>3</sub>Na, 195.0997; found, 195.0990; 5-(4-methoxy-benzyloxy)-4-triethylsilanyloxy-non-1-en-3-ol (**9a**): Pale yellow oil;  $[\alpha]_D^{25} +18.0$  (c 1.0, CHCl<sub>3</sub>); IR (neat):  $v_{max}$ : 3292, 2956, 2871, 1612, 1515, 1251, 1086, 1036 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  7.20 (d, J = 8.3 Hz, 2H), 6.80 (d, J = 9.0 Hz, 2H), 6.00–5.90 (m, 1H), 5.30 (dt, J = 17.3, 1.5 Hz, 1H),

5.18 (dt, J = 10.5, 1.5 Hz, 1H), 4.45 (dd, J = 17.3, 10.5 Hz, 2H), 4.24 (t, J = 6.0 Hz, 1H), 3.80 (s, 3H), 3.74 (t, J = 5.2 Hz, 1H), 3.50-3.42 (m, 1H), 2.20-2.18 (br s, OH, 1H), 1.60-1.30 (m, 4H) 1.00-0.80 (m, 12H), 0.70-0.50 (m, 6H). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): δ 144.7, 142.3, 130.0, 128.7, 128.2, 127.5, 126.2, 122.5, 119.0, 114.8, 75.8, 66.6, 57.4, 45.7, 24.4; MS-ESIMS: m/z 417  $(M+Na)^+$ ; HRMS calcd for  $C_{22}H_{38}O_4NaSi$ , 417.2437; found, 417.2447; 1-(2,2-Dimethyl-5-vinyl-[1,3]dioxolan-4-yl)-butan-1-ol (12): Pale yellow oil;  $[\alpha]_{D}^{25}$  +7.50 (c 1.2, CH<sub>2</sub>Cl<sub>2</sub>); IR (neat): v<sub>max</sub>: 3460, 2930, 1642, 1372, 1252,  $1062 \text{ cm}^{-1}$ ; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>):  $\delta$  6.00 (ddd, J = 17.0, 10.2, 7.2 Hz, 1H), 5.34 (dd, J = 17.0, 10.2 Hz, 2H), 4.58 (t, J = 7.2 Hz, 1H), 3.90 (t, J = 7.2 Hz, 1H), 3.65-3.55 (m, 1H) 1.65-1.55 (m, 2H), 1.50-1.30 (m, 8H), 0.95 (t, J = 7.2 Hz, 3H); <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>):  $\delta$ 135.0, 119.0, 108.5, 80.5, 79.0, 69.5, 35.9, 28.9, 25.3, 18.9,14.0; LC–MS: m/z (%): 223 (M+Na<sup>+</sup>, 10), 200 (M<sup>+</sup>, 5). *Herbarumin-I* (1): colourless solid;  $[\alpha]_D^{25}$  +12.0 (*c* 0.5, EtOH); IR (KBr):  $v_{max}$ : 3431, 2925, 2854, 1630, 1460, 1203, 1090 cm<sup>-1</sup>;<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  5.57 (d, J = 15.6 Hz, 1H), 5.50 (m, 1H), 4.87 (td, J = 9.4, 2.4 Hz, 1H), 4.35 (br s, 1H), 3.41 (d, J = 9.4 Hz, 1H), 2.43 (br s, OH, 1H), 2.22-2.33 (m, 2H), 2.07-1.80 (m, 3H), 1.48-1.70 (m, 3H), 1.30–1.20 (m, 3H), 0.92 (t, *J* = 7.3 Hz, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): 176.0, 130.6, 124.9, 73.7, 73.4, 70.2, 34.4, 33.7, 33.4, 24.7, 18.0, 13.8; MS-LCMS: m/z 251 (M+Na)<sup>+</sup>; HRMS calcd for  $C_{12}H_{20}O_4Na$ , 251.1259; found, 251.1269.